EAST Search History / Interference Search

Ref #	Hits	Search Query	DBs	Default Operat or	Plura Is	Time Stamp
L1	11741	ORGANOSILICON	US-PGPU B; USPAT	ADJ .	OFF	2007/10/24 08:20
L2	957	CYCLIC WITH L1	US-PGPU B; USPAT	ADJ	OFF	2007/10/24 08:20
L3	18	ALKOXYSILANE WITH L2	US-PGPU B; USPAT	ADJ	ON	2007/10/24 08:20

=> d his

(FILE 'HOME' ENTERED AT 07:14:11 ON 24 OCT 2007)

FILE 'REGISTRY' ENTERED AT 07:14:20 ON 24 OCT 2007

L1 STRUCTURE UPLOADED

L2 . 1 S L1

L3 15 S L1 FUL

FILE 'CAPLUS' ENTERED AT 07:15:09 ON 24 OCT 2007

L4 14 S L3

FILE 'REGISTRY' ENTERED AT 07:18:15 ON 24 OCT 2007

L5 STRUCTURE UPLOADED

L6 2016 S L5 FUL

FILE 'CAPLUS' ENTERED AT 07:18:44 ON 24 OCT 2007

FILE 'REGISTRY' ENTERED AT 07:18:49 ON 24 OCT 2007

L7 STRUCTURE UPLOADED

L8 STRUCTURE UPLOADED

L9 74 SEARCH L8 SSS SUB=L6 FULL

FILE 'CAPLUS' ENTERED AT 07:23:00 ON 24 OCT 2007

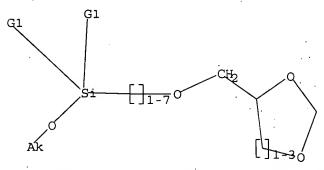
L10 2270 S L9

L11 2 S L10 AND L4

=> d 11

L1 HAS ŅO ANSWERS

L1 STR



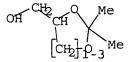
G1 Ak, MeO, EtO, n-PrO, i-PrO, n-BuO, i-BuO, s-BuO, t-BuO

Structure attributes must be viewed using STN Express query preparation.

=> d 18

L8 HAS NO ANSWERS

L8 STR



G1 Ak, MeO, EtO, n-PrO, i-PrO, n-BuO, i-BuO, s-BuO, t-BuO

Structure attributes must be viewed using STN Express query preparation.

=> d bib abs hitstr 1-2 111

L11 ANSWER 1 OF 2 CAPLUS COPYRIGHT 2007 ACS on STN

AN 2005:1074041 CAPLUS

DN 143:369971

TI Sol-gel reaction products, solid electrolytes, protonic conductors, and membrane-electrode assemblies for fuel cells

IN Wariishi, Koji

PA Fuji Photo Film Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 40 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN. CNT 1

PAN.CNI I		'		
PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI JP 2005272556 PRAI JP 2004-86053 OS MARPAT 143:369971	A	20051006 20040324	JP 2004-86053	20040324
GI			•	

$$\begin{bmatrix} R^{1} & O-(CR^{33}R^{34}) & n_{12} \\ C & & & & \\ R^{1} & O-(CR^{31}R^{32}) & n_{11} \end{bmatrix}_{S1} \begin{bmatrix} (R^{5})_{3?m1} \\ \vdots \\ Si-(OR^{6})_{m1} \end{bmatrix}_{L_{1}}$$

The reaction products are prepared from I (R1, R2 = H, alkyl, aryl, heterocyclic ring, R1 and R2 may link together to form a ring; R31, R32, R33, R34, R4 = H, alkyl, aryl, heterocyclic ring; R5 = alkyl, aryl, heterocyclic ring; R6 = H, alkyl, aryl, silyl; m1 = 1-3; n11, n12 = 0-4; L1 = single bond, linkage group with valency (s1 + t1); s1, t1 = 1-4), and compds. having proton-donating substituent groups. The protonic conductors show high protonic conductivity and low methanol permeability.

IT 863015-08-1P

IT 863015-08-1P
 RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT
 (Reactant or reagent)

(preparation of cyclic acetal-containing alkoxysilanes for sol-gel reaction products for protonic conductors of fuel cells)

RN 863015-08-1 CAPLUS

CN Silane, [3-[(2,2-dimethyl-1,3-dioxolan-4-yl)methoxy]propyl]triethoxy-(9CI) (CA INDEX NAME)

Me O
$$CH_2-O-(CH_2)_3-Si-OEt$$
OEt
OEt

IT 100-79-8

RL: RCT (Reactant); RACT (Reactant or reagent) (preparation of cyclic acetal-containing alkoxysilanes for sol-gel reaction products for protonic conductors of fuel cells)

RN 100-79-8 CAPLUS

CN 1,3-Dioxolane-4-methanol, 2,2-dimethyl- (CA INDEX NAME)

IT 863015-08-1DP, polymers with oxidized
mercaptoalkyltrialkoxysilanes and alkoxysilanes 866228-58-2P
RL: DEV (Device component use); IMF (Industrial manufacture); TEM
(Technical or engineered material use); PREP (Preparation); USES (Uses)
(sol-gel reaction products of cyclic acetal-containing alkoxysilanes and proton-donating compds. for protonic conductors of fuel cells)

RN 863015-08-1 CAPLUS

CN Silane, [3-[(2,2-dimethyl-1,3-dioxolan-4-yl)methoxy]propyl]triethoxy-(9CI) (CA INDEX NAME)

RN 866228-58-2 CAPLUS

CN 1-Propanesulfonic acid, 3-(triethoxysilyl)-, polymer with [3-[(2,2-dimethyl-1,3-dioxolan-4-yl)methoxy]propyl]triethoxysilane and 4,4,13,13-tetraethoxy-3,14-dioxa-4,13-disilahexadecane (9CI) (CA INDEX NAME)

CM 1

CRN 863015-08-1 CMF C15 H32 O6 Si

Me O
$$CH_2-O-(CH_2)_3-Si-OEt$$
OEt

CM 2

CRN 260784-99-4 CMF C9 H22 O6 S Si

CM 3

CRN 52217-60-4 CMF C20 H46 O6 Si2

L11 ANSWER 2 OF 2 CAPLUS COPYRIGHT 2007 ACS on STN

AN 2005:902902 CAPLUS

DN 143:230419

TI Preparation of organosilicon compounds for organic silicone resin having diols with good storage stability

IN Komuro, Katsuhiko; Suzuki, Hiroshi

PA Toagosei Co., Ltd., Japan

SO PCT Int: Appl., 24 pp.

CODEN: PIXXD2

DT Patent

LA Japanese

FAN. CNT 1

ΙI

FAN.	$\gamma_{\rm IM,I}$	1																
	PAT	CENT :	NO.			KIN	D	DATE		1	APPL	I CAT	ION I	NO.		D)	ATE	
	-,						-			•								
ΡI	WO	2005	0779	60		A1 20050825		WO 2005-JP1972					20050209					
		W :	ΑE,	AG,	AL,	AM,	ΑT,	AU,	AZ,	BA,	BB,	BG,	BR,	BW,	BY,	BZ,	CA,	CH,
			CN,	CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	EG,	ES,	FI,	GB,	GD,
			GE,	GH,	GM,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	KE,	KG,	KP,	KR,	ΚŻ,	LC,
	•		LK,	LR,	LS,	LT,	LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,	. MX ,	ΜZ,	NA,	NI,
			NO,	NZ,	OM,	PG,	PH,	PL,	PT,	RO,	RU,	SC,	SD,	SE,	SG,	SK,	SL,	SY,
		·	·TJ,	TM,	TN,	TR,	.TT,	TZ,	UA,	UG,	US,	UΖ,	VC,	VN,	YU,	ZA,	ZM,	zw
		RW:	BW,	GH,	GM,	KE,	LS,	MW,	MZ,	NA,	SD,	SL,	SZ,	TZ,	UG,	ZM,	ZW,	AM,
	•		AZ,	ΒY,	KG,	ΚZ,	MD,	RU,	TJ,	TM,	AT,	BE,	BG,	CH,	CY,	CZ,	DE,	DK,
			EE,	ES,	FI,	FR,	GB,	GR,	HU,	ΙE,	IS,	ΙT,	LT,	LU,	MC,	NL,	PL,	PT,
			RO,	SE,	SI,	SK,	TR,	BF,	ВJ,	CF,	CG,	CI,	CM,	GA,	GN,	GQ,	GW,	ML,
			MR,	NE,	SN,	TD,	TG							•				
	KR	2007	0101	24		Α		2007	0122		KR 2	006-	7189	87		2	0060	915
	US	2007	1736	25		A1		2007	0726	1	US 2	007-	5892	62		2	0070	111
PRAI	JР	2004	-390	64		A		2004	0216									
•	JP	2004	-390	65		Α		2004	0216							•		
	WO	2005	-JP1	972		W		2005	0209				• •					
GT			•															

$$R^{5}$$
 O Me $Z-O$ II

AB The organosilicon resin having a diol is prepared by hydrolyzing/condensing an organic compound I obtained by hydrosilylation reaction between a compound

and a silane compound R1R2R3SiH with a polyfunctional alkoxysilane, wherein

R1, R2, R3 = C1-6 alkyl or alkoxy; R4 = C2-6 alkylene; R5 = alkene having a terminal C:C bond and Z = C1-3 alkylene. The organosilicon resin is produced by the hydrolysis/condensation of a mixture comprising the organosilicone compound I and a mol. weight modifier. Thus, 227 mmol 2,2-dimethyl-1,3-dioxolane-4-methanol and 250 mmol allyl bromide were reacted in the presence of sodium hydride, 119 mmol of the resulting 4-[(allyloxy)methyl]-2,2-dimethyl-1,3-dioxolane was reacted with 131 mmol triethoxysilane in the presence of a divinyltetramethyldisiloxane platinum complex to give 4-[(3-triethoxysilylpropyloxy)methyl]-2,2-dimethyl-1,3-dioxolane, 12 mmol of which was reacted with 44.1 mmol methyltrimethoxysilane and 12 mmol hexamethyldisiloxane in the presence of 3.74 g 1.5% aqueous hydrochloric acid for 1.5 h, reacted with hexamethyldisilazane, and dried to give a silsesquioxane resin having diols, showing good storage stability.

IT 863015-08-1P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(intermediate; preparation of organosilicon compds. for organic silicone resin

having diols)

RN 863015-08-1 CAPLUS

CN Silane, [3-[(2,2-dimethyl-1,3-dioxolan-4-yl)methoxy]propyl]triethoxy-(9CI) (CA INDEX NAME)

Me
$$CH_2-O-(CH_2)_3-Si-OEt$$
 OEt
 OEt
 OEt

IT 863015-09-2DP, trimethylsilyl-terminated, deprotected

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(preparation of organosilicon compds. for organic silicone resin having diols)

RN 863015-09-2 CAPLUS

CN Silane, [3-[(2,2-dimethyl-1,3-dioxolan-4-yl)methoxy]propyl]triethoxy-, polymer with trimethoxymethylsilane (9CI) (CA INDEX NAME)

CM 1

CRN 863015-08-1 CMF C15 H32 O6 Si

Me O
$$CH_2-O-(CH_2)_3-Si-OEt$$
 OEt OEt

CM 2

CRN 1185-55-3 CMF C4 H12 O3 Si

IT 100-79-8

RL: RCT (Reactant); RACT (Reactant or reagent) (preparation of organosilicon compds. for organic silicone resin having diols)

RN 100-79-8 CAPLUS

CN 1,3-Dioxolane-4-methanol, 2,2-dimethyl- (CA INDEX NAME)

=>

RE.CNT 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

=> d his

(FILE 'HOME' ENTERED AT 07:14:11 ON 24 OCT 2007)

FILE 'REGISTRY' ENTERED AT 07:14:20 ON 24 OCT 2007

STRUCTURE UPLOADED . L1

1 S L1 L2

L3 15 S L1 FUL

FILE 'CAPLUS' ENTERED AT 07:15:09 ON 24 OCT 2007

L4 14 S L3

FILE 'REGISTRY' ENTERED AT 07:18:15 ON 24 OCT 2007

L5 STRUCTURE UPLOADED

2016 S L5 FUL

FILE 'CAPLUS' ENTERED AT 07:18:44 ON 24 OCT 2007

FILE 'REGISTRY' ENTERED AT 07:18:49 ON 24 OCT 2007

L7 STRUCTURE UPLOADED

L8 STRUCTURE UPLOADED

74 SEARCH L8 SSS SUB=L6 FULL .Ь9

FILE 'CAPLUS' ENTERED AT 07:23:00 ON 24 OCT 2007

2270 S L9 L10

2 S L10 AND L4 L11

SELECT RN L11 2

FILE 'REGISTRY' ENTERED AT 07:31:45 ON 24 OCT 2007

7 S E1-E7

FILE 'REGISTRY' ENTERED AT 07:36:49 ON 24 OCT 2007

STRUCTURE UPLOADED

L1450 S L13

L15 250022 S L13 FUL

FILE 'CAPLUS' ENTERED AT 07:38:03 ON 24 OCT 2007

155778 S L15 L16

14 S L16 AND L4 L17

FILE 'REGISTRY' ENTERED AT 07:40:04 ON 24 OCT 2007

15476 SEARCH L13 CSS SUB=L15 FULL

FILE 'CAPLUS' ENTERED AT 07:40:53 ON 24 OCT 2007

=> s 118

L19 62998 L18 .

=> s 119 and 14

L20 2 L19 AND L4

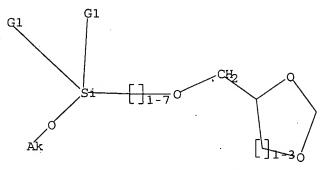
=> d l13

L13 HAS NO ANSWERS

L13. STR

Structure attributes must be viewed using STN Express query preparation.

 $= \dot{d} 11$ L1 HAS NO ANSWERS L1STR



G1 Ak, MeO, EtO, n-PrO, i-PrO, n-BuO, i-BuO, s-BuO, t-BuO

Structure attributes must be viewed using STN Express query preparation.

=> d bib abs hitstr 1-2 120

L20 ANSWER 1 OF 2 CAPLUS COPYRIGHT 2007 ACS.on STN

AN 2007:287048 CAPLUS

DN 146:341046

ΤI Cyclic carbonate-modified organosilicon non-aqueous electrolytes for secondary batteries and capacitors

IN Nakanishi, Tetsuo; Kashida, Meguru; Miyawaki, Satoru

Shin-Estu Chemical Co., Ltd., Japan PA

SO U.S. Pat. Appl. Publ., 12pp.

CODEN: USXXCO

DTPatent

LA English

FAN.CNT 1

			•		
1	PATENT NO.		DATE	APPLICATION NO.	DATE
					,
PI U	US 2007059607 ·	A1 .	20070315	US 2006-514106	20060901
	JP 2007077052	A	20070329	JP 2005-265551	20050913
]	KR 2007030682	Α	20070316	KR 2006-87897	20060912
(CN 1931863	A.	20070321	CN 2006-10151899	20060913

PRAI JP 2005-265551 A 20050913

AB A cyclic carbonate-modified silane or siloxane is combined with a nonaq. solvent and an electrolyte salt to form a non-aqueous electrolytic solution

This

electrolyte is used in a secondary battery which has improved temperature and cycling characteristics.

IT 42345-73-3P

RL: PRP (Properties); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(cyclic carbonate-modified organosilicon non-aqueous electrolytes for secondary batteries and capacitors)

RN 42345-73-3 CAPLUS

CN 1,3-Dioxolan-2-one, 4-[[3-(trimethoxysilyl)propoxy]methyl]- (CA INDEX NAME)

IT 1825-61-2, Trimethylmethoxysilane

RL: RCT (Reactant); RACT (Reactant or reagent)

(cyclic carbonate-modified organosilicon non-aqueous electrolytes for secondary batteries and capacitors)

RN 1825-61-2 CAPLUS

CN Silane, methoxytrimethyl- (CA INDEX NAME)

L20 ANSWER 2 OF 2 CAPLUS COPYRIGHT 2007 ACS on STN

AN 2005:902902 CAPLUS

DN 143:230419

TI Preparation of organosilicon compounds for organic silicone resin having diols with good storage stability

IN Komuro, Katsuhiko; Suzuki, Hiroshi

PA Toagosei Co., Ltd., Japan

SO . PCT Int. Appl., 24 pp.

CODEN: PIXXD2

DT Patent

LA Japanese

FAN CNT 1

ran.cni i																
	PATENT NO		K	IND	DATE		. 1	APPL:	ICAT:	I NO I	NO.		D?	ATE		
			_													
ΡI	PI WO 2005077960			A1	2005	0825	7	NO 2	005-0	JP19'	72		20050209 BY, BZ, CA, CH			
	W: Al	E, AG,	AL, A	м, АТ,	AU,	AZ,	BA,	BB,	BG,	BR,	BW,	BY,	BZ,	CA,	CH,	
	. Ci	1, CO,	CR, C	U, CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	EG,	ES,	FI,	GB,	GD,	
	GI	Ξ, GH,	GM, H	R, HU,	ID,	ΙL,	IN,	IS,	JP,	KE,	KG,	ΚP,	KR,	KZ,	LC,	
	Li	K, LR,	LS, L	Γ, LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,	MX,	MZ,	NA,	NI,	
	NO	O, NZ,	OM, P	G, PH,	PL,	PT,	RO,	RU,	SC,	SD,	SE,	SG,	SĶ,	SL,	SY,	
	To	J, TM,	TN, T	R, TT,	TZ,	UA,	UG,	US,	UZ,	VC,	VN,	YU,	ZA,	ZM,	ZW	
	RW: BY	V. GH.	GM. K	E LS	MW	M7.	NΑ	SD	ST.	S7.	TZ	HG	7.M	7W	ΔM	

AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG KR 2006-718987 KR 2007010124 Α 20070122 20060915 US 2007173625 A1 20070726 US 2007-589262 20070111 PRAI JP 2004-39064 Α 20040216 JP 2004-39065 Α 20040216 WO 2005-JP1972 W 20050209 GI

AB The organosilicon resin having a diol is prepared by hydrolyzing/condensing an organic compound I obtained by hydrosilylation reaction between a compound

and a silane compound R1R2R3SiH with a polyfunctional alkoxysilane, wherein R1, R2, R3 = C1-6 alkyl or alkoxy; R4 = C2-6 alkylene; R5 = alkene having a terminal C:C bond and Z = C1-3 alkylene. The organosilicon resin is produced by the hydrolysis/condensation of a mixture comprising the organosilicone compound I and a mol. weight modifier. Thus, 227 mmol 2,2-dimethyl-1,3-dioxolane-4-methanol and 250 mmol allyl bromide were reacted in the presence of sodium hydride, 119 mmol of the resulting 4-[(allyloxy)methyl]-2,2-dimethyl-1,3-dioxolane was reacted with 131 mmol triethoxysilane in the presence of a divinyltetramethyldisiloxane platinum complex to give 4-[(3-triethoxysilylpropyloxy)methyl]-2,2-dimethyl-1,3-dioxolane, 12 mmol of which was reacted with 44.1 mmol methyltrimethoxysilane and 12 mmol hexamethyldisiloxane in the presence of 3.74 g 1.5% aqueous hydrochloric acid for 1.5 h, reacted with hexamethyldisilazane, and dried to give a silsesquioxane resin having diols, showing good storage stability.

IT 863015-08-1P

ΙI

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(intermediate; preparation of organosilicon compds. for organic silicone resin

having diols)

RN 863015-08-1 CAPLUS

CN Silane, [3-[(2,2-dimethyl-1,3-dioxolan-4-yl)methoxy]propyl]triethoxy-(9CI) (CA INDEX NAME)

IT 863015-09-2DP, trimethylsilyl-terminated, deprotected

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(preparation of organosilicon compds. for organic silicone resin having diols)

RN 863015-09-2 CAPLUS

CN Silane, [3-[(2,2-dimethyl-1,3-dioxolan-4-yl)methoxy]propyl]triethoxy-, polymer with trimethoxymethylsilane (9CI) (CA INDEX NAME)

CM 1

CRN 863015-08-1 CMF C15 H32 O6 Si

Me O
$$CH_2-O-(CH_2)_3-Si-OEt$$

Me O $CH_2-O-(CH_2)_3-Si-OEt$

OEt

CM 2

CRN 1185-55-3 CMF C4 H12 O3 Si

=>

RE.CNT 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

=> d his

(FILE 'HOME' ENTERED AT 07:14:11 ON 24 OCT 2007)

FILE 'REGISTRY' ENTERED AT 07:14:20 ON 24 OCT 2007

L1 STRUCTURE UPLOADED

L2 1 S L1

L3 15 S L1 FUL

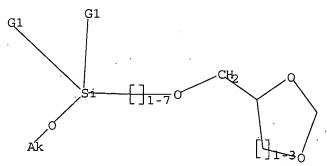
FILE 'CAPLUS' ENTERED AT 07:15:09 ON 24 OCT 2007

L4 14 S L3

=> d 11

L1 HAS NO ANSWERS

L1 STR



G1 Ak, MeO, EtO, n-PrO, i-PrO, n-BuO, i-BuO, s-BuO, t-BuO

Structure attributes must be viewed using STN Express query preparation.

=> d bib abs hitstr 1-14

L4 ANSWER 1 OF 14 CAPLUS COPYRIGHT 2007 ACS on STN

AN 2007:789423 CAPLUS

DN 147:235149

TI Method for manufacturing cyclic carbonate from epoxy compound and carbon dioxide

IN Hua, Ruimao; Jiang, Jiali; Ma, Deqiang; Song, Jinhong; Shang, Yonghua; Ding, Jiansheng

PA Ningbo Wanhua Polyurethane Co., Ltd., Peop. Rep. China

SO Faming Zhuanli Shenqing Gongkai Shuomingshu, 10pp. CODEN: CNXXEV

DT Patent

LA Chinese

FAN.CNT 1

PATE	NT NO.	KIND	DATE	APPLICATION NO.	DATE
	00999514 006-10170993	Α	20070718 20061228	CN 2006-10170993	20061228

OS CASREACT 147:235149

AB In the invention, the catalyst system is a homogeneous catalyst system with amide RCONR1R2 (R = H, Me, Et; R1, R2 = H, Me, Et, n-Pr, iso-Pr, Bu, etc.) as a catalyst and water as a promoter. The title method comprises the steps of: (1) adding amide to a reactor, adding water and epoxy compound, and sealing, (2) pumping carbon dioxide gas into the reactor, stirring, heating and reacting, and (3) stopping heating after reacting for 1-20 h, stirring, cooling to room temperature, and releasing the gas. The method has the advantages of low catalyst cost, high catalyst activity and

high reaction efficiency. The catalyst is stable, environment-friendly, and easy to reuse.

IT 42345-73-3P

RL: IMF (Industrial manufacture); SPN (Synthetic preparation); PREP (Preparation)

(preparation of cyclic carbonate by addition reaction of epoxide with carbon dioxide in presence of amide and water)

RN 42345-73-3 CAPLUS

CN 1,3-Dioxolan-2-one, 4-[[3-(trimethoxysily1)propoxy]methyl]- (CA INDEX NAME)

$$OMe$$
 $OH_2-O-(CH_2)_3-Si-OMe$
 OMe
 OMe
 OMe

L4 ANSWER 2 OF 14 CAPLUS COPYRIGHT 2007 ACS on STN

AN 2007:287048 CAPLUS

DN 146:341046

TI Cyclic carbonate-modified organosilicon non-aqueous electrolytes for secondary batteries and capacitors

IN Nakanishi, Tetsuo; Kashida, Meguru; Miyawaki, Satoru

PA Shin-Estu Chemical Co., Ltd., Japan

SO U.S. Pat. Appl. Publ., 12pp.

CODEN: USXXCO

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 2007059607	 A1	20070315	US 2006-514106	20060901
	JP 2007077052	A	20070329	JP 2005-265551	20050913
	KR 2007030682	Α.	20070316	KR 2006-87897	20060912
	CN 1931863	A	20070321	CN 2006-10151899	20060913
PRAI	JP 2005-265551	A	20050913		

AB A cyclic carbonate-modified silane or siloxane is combined with a nonaq. solvent and an electrolyte salt to form a non-aqueous electrolytic solution This

electrolyte is used in a secondary battery which has improved temperature and cycling characteristics.

IT 42345-73-3P

RL: PRP (Properties); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(cyclic carbonate-modified organosilicon non-aqueous electrolytes for secondary batteries and capacitors)

RN 42345-73-3 CAPLUS

CN 1,3-Dioxolan-2-one, 4-[[3-(trimethoxysilyl)propoxy]methyl]- (CA INDEX NAME)

$$O$$
 $CH_2-O-(CH_2)_3-Si-OMe$ OMe OMe OMe

L4 ANSWER 3 OF 14 CAPLUS COPYRIGHT 2007 ACS on STN

AN 2006:1198474 CAPLUS

DN 146:121863

TI Efficient DMF-catalyzed coupling of epoxides with CO2 under solvent-free conditions to afford cyclic carbonates

AU Jiang, Jia-Li; Hua, Ruimao

CS Department of Chemistry, Innovative Catalysis Program, Key Laboratory of Organic Optoelectronics and Molecular Engineering of Ministry of Education, Tsinghua University, Beijing, Peop. Rep. China

SO Synthetic Communications (2006), 36(21), 3141-3148 CODEN: SYNCAV; ISSN: 0039-7911

PB Taylor & Francis, Inc.

DT Journal

LA English

OS CASREACT 146:121863

AB To develop a simple, low-mol., and cost-effective organocatalyst for the coupling of epoxides with CO2, we have screened this coupling reaction in different organic solvents and found that DMF is an efficient organic catalyst for the coupling of epoxides with CO2 to give cyclic carbonates in high yield. In some cases, the catalytic activity of DMF can be significantly increased by the addition of catalytic amount of H2O.

IT 42345-73-3P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of cyclic carbonates by DMF-catalyzed coupling of epoxides with carbon dioxide under solvent-free conditions)

RN 42345-73-3 CAPLUS

CN 1,3-Dioxolan-2-one, 4-[[3-(trimethoxysily1)propoxy]methy1]- (CA INDEX NAME)

$$\begin{array}{c} \text{OMe} \\ \text{CH}_2\text{-O-(CH}_2)_3\text{-}\text{Si-OMe} \\ \text{OMe} \\ \text{OMe} \end{array}$$

RE.CNT 44 THERE ARE 44 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L4 ANSWER 4 OF 14 CAPLUS COPYRIGHT 2007 ACS on STN

AN 2005:1074041 CAPLUS

DN 143:369971

TI Sol-gel reaction products, solid electrolytes, protonic conductors, and membrane-electrode assemblies for fuel cells

IN Wariishi, Koji

PA Fuji Photo Film Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 40 pp.

CODEN: JKXXAF

DT Patent

LA Japanese ·

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE	
ΡI	JP 2005272556	Α	20051006	JP 2004-86053	20040324	
PRAI	JP 2004-86053		20040324			
OS	MARPAT 143:369971			•		

GΙ

$$\begin{bmatrix} R^{1} & O-(CR^{33}R^{34}) & n_{12} \\ C & CR^{4} & L_{1} & Si-(OR^{6}) & m_{1} \\ R^{1} & O-(CR^{31}R^{32}) & n_{11} \end{bmatrix}_{S_{1}}$$

The reaction products are prepared from I (R1, R2 = H, alkyl, aryl, heterocyclic ring, R1 and R2 may link together to form a ring; R31, R32, R33, R34, R4 = H, alkyl, aryl, heterocyclic ring; R5 = alkyl, aryl, heterocyclic ring; R6 = H, alkyl, aryl, silyl; m1 = 1-3; n11, n12 = 0-4; L1 = single bond, linkage group with valency (s1 + t1); s1, t1 = 1-4), and compds. having proton-donating substituent groups. The protonic conductors show high protonic conductivity and low methanol permeability. IT 863015-08-1P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(preparation of cyclic acetal-containing alkoxysilanes for sol-gel reaction products for protonic conductors of fuel cells)

RN 863015-08-1 CAPLUS

CN Silane, [3-[(2,2-dimethyl-1,3-dioxolan-4-yl)methoxy]propyl]triethoxy-(9CI) (CA INDEX NAME)

Me O
$$CH_2-O-(CH_2)_3-Si-OEt$$
Ne O $CH_2-O-(CH_2)_3-Si-OEt$

CN· Silane, [3-[(2,2-dimethyl-1,3-dioxolan-4-yl)methoxy]propyl]triethoxy-(9CI) (CA INDEX NAME)

Me O
$$CH_2-O-(CH_2)_{,3}-S_{i}-OEt$$
OEt
OEt

RN 866228-58-2 CAPLUS

1-Propanesulfonic acid, 3-(triethoxysily1)-, polymer with

[3-[(2,2-dimethyl-1,3-dioxolan-4-yl)methoxy]propyl]triethoxysilane and
4,4,13,13-tetraethoxy-3,14-dioxa-4,13-disilahexadecane (9CI) (CA INDEX NAME)

CM 1

CRN 863015-08-1 CMF C15 H32 O6 Si

Me O
$$CH_2-O-(CH_2)_3-Si-OEt$$
OEt
OEt

CM 2

CRN 260784-99-4 CMF C9 H22 O6 S Si

$$\begin{array}{c} \text{OEt} \\ \mid \\ \text{EtO-Si-} (\text{CH}_2)_3 - \text{SO}_3\text{H} \\ \mid \\ \text{OEt} \end{array}$$

CM 3

CRN 52217-60-4 CMF C20 H46 O6 Si2

```
L4 ANSWER 5 OF 14 CAPLUS COPYRIGHT 2007 ACS on STN
```

AN 2005:902902 CAPLUS

DN 143:230419

TI Preparation of organosilicon compounds for organic silicone resin having diols with good storage stability

IN Komuro, Katsuhiko; Suzuki, Hiroshi

PA Toagosei Co., Ltd., Japan

SO PCT Int. Appl., 24 pp.

CODEN: PIXXD2

DT Patent

LA Japanese

FAN.CNT 1

FAN.CNI I																
	PATENT I	NO.		KIN	D	DATE	•	1	APPL	I CAT	ION	NO.		D	ATE	
			-		-									_		
PI	WO 2005	077960		A1		2005	0825	Ţ	WO 2	005-	JP19	72		20	0050	209
	W:	AE, AG	, AL,	AM,	AT,	AU,	ΑZ,	BA,	BB,	BG,	BR,	BW,	BY,	ΒZ,	CA,	CH,
		CN, CO	, CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	EG,	ES,	FI,	GB,	GD,
	•	GE, GH	, GM,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	KE,	KG,	KP,	KR,	KZ,	LC,
		LK, LR	, LS,	LT,	LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,	MX,	MZ,	NA,	NI,
		NO, NZ	, OM,	PG,	PH,	PL,	PT,	·RO,	RU,	SC,	SD,	SE,	SG,	SK,	SL,	SY,
		TJ, TM	, TN,	TR,	TT,	TZ,	UA,	UG,	US,	UZ,	VC,	VN,	YU,	ZA,	ZM,	ZW
	RW:	BW, GH	, GM,	ΚE,	LS,	MW,	MZ,	NA,	SD,	SL,	SZ,	TZ,	UG,	ZM,	ZW,	AM,
		AZ, BY	, KG,	KZ,	MD,	RU,	TJ,	TM,	AT,	BE,	BG,	CH,	CY,	CZ,	DE,	DK,

EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG KR 2006-718987 20060915 KR 2007010124 20070122 Α US 2007173625 US 2007-589262 20070111 A1 20070726 PRAI JP 2004-39064 Α 20040216 JP 2004-39065 20040216 Α WO 2005-JP1972 W 20050209 GI

AB The organosilicon resin having a diol is prepared by hydrolyzing/condensing an organic compound I obtained by hydrosilylation reaction between a compound II

and a silane compound R1R2R3SiH with a polyfunctional alkoxysilane, wherein R1, R2, R3 = C1-6 alkyl or alkoxy; R4 = C2-6 alkylene; R5 = alkene having a terminal C:C bond and Z = C1-3 alkylene. The organosilicon resin is produced by the hydrolysis/condensation of a mixture comprising the organosilicone compound I and a mol. weight modifier. Thus, 227 mmol 2,2-dimethyl-1,3-dioxolane-4-methanol and 250 mmol allyl bromide were reacted in the presence of sodium hydride, 119 mmol of the resulting 4-[(allyloxy)methyl]-2,2-dimethyl-1,3-dioxolane was reacted with 131 mmol triethoxysilane in the presence of a divinyltetramethyldisiloxane platinum complex to give <math>4-[(3-triethoxysilylpropyloxy)methyl]-2,2-dimethyl-1,3-dioxolane, 12 mmol of which was reacted with 44.1 mmol methyltrimethoxysilane and 12 mmol hexamethyldisiloxane in the presence of 3.74 g 1.5% aqueous hydrochloric acid for 1.5 h, reacted with hexamethyldisilazane, and dried to give a silsesquioxane resin having diols, showing good storage stability.

IT 863015-08-1P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(intermediate; preparation of organosilicon compds. for organic silicone resin

having diols)

RN 863015-08-1 CAPLUS

CN Silane, [3-[(2,2-dimethyl-1,3-dioxolan-4-yl)methoxy]propyl]triethoxy-(9CI) (CA INDEX NAME)

Me O
$$CH_2-O-(CH_2)_3-Si-OEt$$
OEt

IT 863015-09-2DP, trimethylsilyl-terminated, deprotected RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

 $\hbox{(preparation of organosilicon compds. for organic silicone resin having diols)} \\$

RN 863015-09-2 CAPLUS

CN Silane, [3-[(2,2-dimethyl-1,3-dioxolan-4-yl)methoxy]propyl]triethoxy-, polymer with trimethoxymethylsilane (9CI) (CA INDEX NAME)

CM 1

CRN 863015-08-1 CMF C15 H32 O6 Si

Me O
$$CH_2$$
-O- $(CH_2)_3$ -Si-OEt OEt OEt

CM 2

CRN 1185-55-3 CMF C4 H12 O3 Si

RE.CNT 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L4 ANSWER 6 OF 14 CAPLUS COPYRIGHT 2007 ACS on STN

AN 2003:211150 CAPLUS

DN 139:7268

TI. Synthesis and dielectric constants of polymers with cyclic carbonate pendant groups

AU Purdy, Andrew P.; Levien, Elizabeth; Hwang, Ann

CS Chemistry Division, Naval Research Laboratory, Washington, DC, 20375-5342, USA

SO Polymer Preprints (American Chemical Society, Division of Polymer Chemistry) (2003), 44(1), 854-855 CODEN: ACPPAY; ISSN: 0032-3934

PB American Chemical Society, Division of Polymer Chemistry

DT Journal; (computer optical disk)

LA English

Polymers containing pendant 5-membered cyclic carbonate functionalities were prepared and their dielec. consts. were measured as a function of frequency. Glycerol carbonate methacrylate, a known compound, was polymerized in bulk. It had a dielec. constant-6 at 1 kHz which dropped to ~5 at 1 MHz, with an dielec. loss ~ 0.1 at 1 kHz. Copolymers with Me methacrylate had lower dielec. consts., with similar loss factors. A silicone polymer with propoxy-glycerol carbonate pendant groups was also prepared and crosslinked with varying amts. of Jeffamine T-403 or triethylenetetramine. Dielec. consts. >20 at 1 kHz were obtained, but the materials had high dielec. losses, and showed some ionic conductivity that could have come from autoionization of the hydroxyethyl carbamate crosslink moieties.

IT 532947-06-1P
 RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
 (Reactant or reagent)

(monomer; synthesis and dielec. consts. of polymers with cyclic carbonate pendant groups)

RN 532947-06-1 CAPLUS

CN 1,3-Dioxolan-2-one, 4-[[3-(diethoxymethylsilyl)propoxy]methyl]- (CA INDEX NAME)

IT 532947-10-7DP, trimethylsilyl-terminated

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(oligomeric; synthesis and dielec. consts. of polymers with cyclic carbonate pendant groups)

RN 532947-10-7 CAPLUS

CN 1,3-Dioxolan-2-one, 4-[[3-(diethoxymethylsily1)propoxy]methyl]-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 532947-06-1 CMF C12 H24 O6 Si

IT 532947-13-0P 532947-15-2P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (synthesis and dielec. consts. of polymers with cyclic carbonate pendant groups)

RN 532947-13-0 CAPLUS

CN 1,3-Dioxolan-2-one, 4-[[3-(diethoxymethylsilyl)propoxy]methyl]-, polymer with N,N'-bis(2-aminoethyl)-1,2-ethanediamine (9CI) (CA INDEX NAME)

CM 1

CRN 532947-06-1 CMF C12 H24 O6 Si

OEt

$$CH_2-O-(CH_2)_3-Si-Me$$

OEt

OEt

CM 2

CRN 112-24-3

CMF C6 H18 N4

 $H_2N-CH_2-CH_2-NH-CH_2-CH_2-NH-CH_2-CH_2-NH_2$

RN 532947-15-2 CAPLUS

CN 1,3-Dioxolan-2-one, 4-[[3-(diethoxymethylsilyl)propoxy]methyl]-, polymer with α -hydro- ω -(2-aminomethylethoxy)[poly[oxy(methyl-1,2-ethanediyl)]] ether with 2-ethyl-2-(hydroxymethyl)-1,3-propanediol (3:1) (9CI) (CA INDEX NAME)

CM 1

CRN 532947-06-1 CMF C12 H24 O6 Si

CM 2

CRN 39423-51-3

CMF (C3 H6 O)n (C3 H6 O)n (C3 H6 O)n C15 H35 N3 O3

CCI. IDS, PMS

PAGE 1-A

$$CH_2$$
—
 CH_2 —
 CH

PAGE 1-B

3 (D1-Me)

$$-(C_3H_6)$$
 $-\frac{1}{n}$ $0-CH_2-CH_2-NH_2$

$$-(C_3H_6)$$
 $- \int_n$ $O-CH_2-CH_2-NH_2$

RE.CNT 11 THERE ARE 11 CITED REFERENCES AVAILABLE FOR THIS RECORD

ALL CITATIONS AVAILABLE IN THE RE FORMAT

L4 ANSWER 7 OF 14 CAPLUS COPYRIGHT 2007 ACS on STN

AN 1998:335569 CAPLUS

DN 129:67867

TI Preparation of 1,3-dioxolane-ring-containing alkoxysilanes as silane coupling agents

IN Ishikawa, Kazunori

PA Yokohama Rubber Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 6 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

PATENT NO.		KIND	DATE	DATE		
PI,	JP 10139787	A	19980526	JP 1996-294933	19961107	
	JP 3819087	B2	20060906			
PRAI	JP 1996-294933	• .	19961107			

OS MARPAT 129:67867

AB Compds. containing 1,3-dioxolane ring and alkoxysilyl group are prepared as silane coupling agents for adhesives, sealants, and primers (no data). A 187, an epoxy silane, was treated with BF3.0Et2 in Me2CO to give 52% 2,2-dimethyl-4-(3-trimethoxysilylpropoxy)methyl-1,3-dioxolane.

IT 208923-72-2P

RL: SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(preparation of dioxolane-containing alkoxysilanes as silane coupling agents)

RN 208923-72-2 CAPLUS

CN Silane, [3-[(2,2-dimethyl-1,3-dioxolan-4-yl)methoxy]propyl]trimethoxy-(9CI) (CA INDEX NAME)

Me O
$$CH_2-O-(CH_2)_3-Si-OMe$$
 OMe OMe

- L4 ANSWER 8 OF 14 CAPLUS COPYRIGHT 2007 ACS on STN
- AN 1996:171857 CAPLUS
- DN 124:203302
- TI Hydrolyzable and polymerizable silanes containing 2-oxo-1,3-dioxacycloalkyl groups and their preparation and use
- IN Wolter, Herbert
- PA Fraunhofer-Gesellschaft zur Foerderung der Angewandten Forschung e.V., Germany
- SO Ger., 30 pp. CODEN: GWXXAW
- DT Patent
- LA German

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PΙ	DE 4423811	C1	19960118	DE 1994-4423811	19940706
	EP 694550	A2	19960131	EP 1995-107099	19950511
	EP 694550	A3	19980415		
	EP 694550	B1	20010808	,	

	R: AT, BE, CH	, DE,	OK, FR, GB, IT, LI, LU, N	L, SE
	AT 203995	· . T	20010815 AT 1995-10	7099 19950511
	US 5756767	Α	19980526 US 1997-85	6939 19970515
	US 5917075	A	19990629 US 1997-99	5373 19971222
PRAI	DE 1994-4423811	A	19940706	
	US 1995-499026	B1	19950706	
	US 1997-856939	A3	19970515	
OS	MARPAT 124:203302			
AB	The title compds ,	e.g.,	RCH2O(CH2)3Si(OMe)2R1 (I	; R =

The title compds., e.g., RCH2O(CH2)3Si(OMe)2R1 (I; R = 2-oxo-1,3-dioxacyclopentyl; R1 = OMe, Me), are prepared for use as monomers and reactants. Reacting (3-glycidyloxypropyl)trimethoxysilane with CO2 gave I (R1 = OMe) which was reacted with H2NCH2CH2NH2 to give [(MeO)3Si(CH2)3OCH2CH(OH)CH2O2CNHCH2]2.

IT 42345-73-3P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(preparation as hydrolyzable and polymerizable monomer and reactant and ring-opening reaction with ethylenediamine)

RN 42345-73-3 CAPLUS

CN 1,3-Dioxolan-2-one, 4-[[3-(trimethoxysily1)propoxy]methyl]- (CA INDEX NAME)

$$\begin{array}{c} \text{OMe} \\ | \\ \text{CH}_2\text{-O-(CH}_2)}_3\text{-Si-OMe} \\ | \\ \text{OMe} \end{array}$$

IT 174569-12-1P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(preparation of hydrolyzable and polymerizable)

RN 174569-12-1 CAPLUS

CN 1,3-Dioxolan-2-one, 4-[[3-(dimethoxymethylsilyl)propoxy]methyl]- (9CI) (CA INDEX NAME)

OMe

$$CH_2-O-(CH_2)_3-Si-Me$$

OMe

OMe

OMe

L4 ANSWER 9 OF 14 CAPLUS COPYRIGHT 2007 ACS on STN

AN 1995:222363 CAPLUS

DN 122:291655

TI A new silane precursor with reduced polymerization shrinkage

AU Wolter, H.; Storch, W.

CS Fraunhofer-Institut Silicatforschung, Wuerzburg, Germany

SO Journal of Sol-Gel Science and Technology (1994), 2(1/2/3), 93-6 CODEN: JSGTEC; ISSN: 0928-0707

PB Kluwer

DT Journal

LA English

AB Silsesquioxanes with reduced shrinkage during curing are prepared from 2-[[3-(trimethoxysilyl)propoxy]methyl]-1,4,6-trioxaspiro[4.4]nonane (I). The I is prepared by condensation of butyrolactone with [3-

(glycidyloxy)propyl]trimethoxysilane. The polymer network is formed by hydrolytic polymerization of the trimethoxysilane moiety and the crosslinking

is carried out by a thermal or photochem. cationic ring-opening process.

ΙT 148887-01-8P, 2-[[3-(Trimethoxysilyl)propoxy]methyl]-1,4,6-

trioxaspiro[4.4]nonane

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(monomer; in preparation of low-shrinkage curing silsesquioxanes)

148887-01-8 CAPLUS RN

Silane, trimethoxy[3-(1,4,6-trioxaspiro[4.4]non-2-ylmethoxy)propyl]- (9CI) CN (CA INDEX NAME)

$$CH_2-O-(CH_2)_3-Si-OMe$$
OMe
OMe

163214-76-4P, 2-[[3-(Trimethoxysily1)propoxy]methyl]-1,4,6-IT

trioxaspiro[4.4] nonane homopolymer

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of low-shrinkage curing)

RN 163214-76-4 CAPLUS

Silane, trimethoxy[3-(1,4,6-trioxaspiro[4.4]non-2-ylmethoxy)propyl]-, CN homopolymer (9CI) (CA INDEX NAME)

CM1

CRN 148887-01-8 CMF C13 H26 O7 Si

$$CH_2-O-(CH_2)_3-Si-OMe$$
OMe
OMe

ANSWER 10 OF 14 CAPLUS COPYRIGHT 2007 ACS on STN L4

AN 1993:473279 CAPLUS

DN119:73279

Preparation and use of polymerizable and hydrolyzable silanes TI

IN Wolter, Herbert

PA Fraunhofer-Gesellschaft zur Foerderung der Angewandten Forschung eV, Germany

SO Ger., 20 pp.

CODEN: GWXXAW

DT Patent

T.A German

FAN CNT 1

1711	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
	_ =				
PI	DE 4125201	C1	19921001	DE 1991-4125201	19910730
	EP 525392	A1	19930203	EP 1992-110752	19920625

```
EP 525392
                           В1
                                 19960821
         R: AT, BE, CH, DE, DK, FR, GB, IT, LI, LU, NL, SE
                                 19960915
                                             AT 1992-110752
    AT 141607
                           Т
                                                                      19920625
     US 5414093
                           Α
                                 19950509
                                             US 1992÷916584
                                                                      19920720
     JP 05222199
                           Α
                                 19930831
                                              JP 1992-204107
                                                                      19920730
     JP 3187150
                           B2
                                 20010711
PRAI DE 1991-4125201
                           Α
                                 19910730
    MARPAT 119:73279
OS
```

AB The silanes YnSiXmR4-m-n (X = H, halogen, OH, alkoxy, acyloxy, acyl, carboalkoxy, amino; Y = 1,4,6-trioxaspiro[4.4]nonyl group, optionally substituted; m, n = 1-3; m + n ≤4), useful in the preparation of coatings, adhesives, sealants, etc., are prepared Adding 307 g [3-(glycidyloxy)propyl]trimethoxysilane in 300 mL CH2Cl2 over 1 h to 129 g γ-butyrolactone and 4.62 g BF3.Et2O in 600 mL CH2Cl2 stirred at room temperature and stirring for 2 h gave 2-[[3-(trimethoxysilyl)propoxy]methyl]-1,4,6-trioxaspiro[4.4]nonane (I). Adding 1.20 mg Et3N and 0.54 g H2O dropwise to 6.54 g I and stirring for apprx.20 h at room temperature gave a siloxane which could be cured by cationic polymerization

IT 148887-01-8P

RL: RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent) (preparation and hydrolytic polymerization of)

RN 148887-01-8 CAPLUS

CN Silane, trimethoxy[3-(1,4,6-trioxaspiro[4.4]non-2-ylmethoxy)propyl]- (9CI) (CA INDEX NAME)

$$CH_2-O-(CH_2)_3-Si-OMe$$
OMe
OMe

L4 ANSWER 11 OF 14 CAPLUS COPYRIGHT 2007 ACS on STN

AN 1986:609334 CAPLUS

DN 105:209334

TI 1-0-[3-(Triethoxysily1)propy1] - and 1-0-[3-(diethoxymethylsily1)propy1] - 2,3:4,5-di-0-isopropylidene-D-arabinitol

IN Capka, Martin; Hetflejs, Jiri; Holy, Antonin; Rosenberg, Ivan

PA Czech.

SO Czech., 3 pp.

CODEN: CZXXA9

DT Patent

LA Czech

FAN.CNT 1

	PA.	FENT NO.	KIND	DATE	APPLICATION NO.	DATE
ΡI	CS	229088	B1	19840514	CS 1982-8567 .	19821129
PRAI	CS	1982-8567	•	19821129		

GI For diagram(s), see printed CA Issue.

AB The title compds. (I; R = OEt, Me) were prepared in 41-45% yield by addition of (EtO)3SiH and (EtO)2MeSiH, resp., to 1-O-allyl-2,3:4,5-di-O-isopropylidene-D-arabinitol at 120° in THF, catalyzed by H2PtCl6. I are modifiers of inorg. supports in chromatog. and immobilization of biol. substances.

IT 105239-74-5P 105239-75-6P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of, as chromatog. support modifier, for immobilization of biol. substances)

RN 105239-74-5 CAPLUS

CN D-Arabinitol, 2,3:4,5-bis-O-(1-methylethylidene)-1-O-[3-(triethoxysilyl)propyl] - (9CI) (CA INDEX NAME)

Me O Me OEt
$$CH_2-O-(CH_2)_3-Si-OEt$$
 OEt

RN 105239-75-6 CAPLUS

CND-Arabinitol, 1-O-[3-(diethoxymethylsilyl)propyl]-2,3:4,5-bis-O-(1methylethylidene) - (9CI) (CA INDEX NAME)

L4ANSWER 12 OF 14 CAPLUS COPYRIGHT 2007 ACS on STN

AN 1979:205196 CAPLUS

DN 90:205196

TI Use of silanes having capped functional groups as adhesivizing agents

IN Amort, Juergen; Nestler, Heinz

Dynamit Nobel A.-G., Fed. Rep. Ger. PA

SO U.S., 5 pp.

CODEN: USXXAM

DT Patent

LΑ English

FAN. CNT 2									
	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE				
ΡI	US 4118540	A	19781003	US 1976-755876	19761230				
	DE 2559259	A1	19770714	DE 1975-2559259	19751231				
PRAI	DE 1975-2559259	Α	19751231						
AB	Compds. such as (Et	O) 2CHCH	2NH(CH2)3Si	(OMe)3 [63968-38-7] an	d .				
	4-[[3-(trimethoxysilyl)propoxy]methyl]-1,3-dioxolane [50650-15-2								
] are useful as coupling agents for improving the adhesion between glass								
	fibers, sand, metal surfaces, etc., and resins such as epoxy and furan								
	resins. Mixts of the coupling agents and resins, especially phenolic resol								
	resins, have a long shelf life.								
ΙT	50650-15-2								
	RL: USES (Uses)								
	(coupling agents	, for i	norg. oxide	and metal surfaces and	resins)				
		•	_						

RN 50650-15-2 CAPLUS

CN Silane, [3-(1,3-dioxolan-4-ylmethoxy)propyl]trimethoxy- (9CI) (CA INDEX NAME)

L4 ANSWER 13 OF 14 CAPLUS COPYRIGHT 2007 ACS on STN

Ι

AN 1977:519455 CAPLUS

DN 87:119455

TI Silanes with blocked functional groups as adhesion promoters

IN Nestler, Heinz; Amort, Juergen

PA Dynamit Nobel A.-G., Fed. Rep. Ger.

SO Ger. Offen., 12 pp.

CODEN: GWXXBX

DT Patent

LA German

FAN.CNT 2

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
		<i></i>			
ΡI	DE 2559259	A1	19770714	DE 1975-2559259	19751231
	GB 1577928	A	19801029	GB 1976-52949	19761217
	GB 1577929	. A	19801029	GB 1979-21916	19761217
	JP 52084240	A	19770713	JP 1976-160781	19761228
	NL 7614579	A	19770704	NL 1976-14579	19761230
	FR 2336983	A1	19770729	FR 1976-39630	19761230
	FR 2336983	B1	19821203		
	ÚS 4118540	A	19781003 .	US 1976-755876	19761230
PRAI	DE 1975-2559259	A	19751231		
	GB 1976-52949	Α	19761217	•	
CT	·				

GI

- AB Alkoxysilanes containing acetal groups or 1,3-doxolane derivs. are useful as corrosion-protective coatings for metals or as couplers for improved adhesion. Thus, a degreased 15 + 8-cm Cu plate is immersed in a 10% alc. HOAc solution of silane I [50650-15-2] and dried 1 h at 130° to give a hard, adherent film which cannot be detached by scratching with a knife.
- IT 50650-15-2

RL: USES (Uses)

(coatings for metals and couplers for improved adhesion in fiber-reinforced plastics)

RN 50650-15-2 CAPLUS

CN Silane, [3-(1,3-dioxolan-4-ylmethoxy)propyl]trimethoxy- (9CI) (CA INDEX NAME)

$$\begin{array}{c} \text{OMe} \\ | \\ \text{O} \\ \text{OH2} \\ \text{O} \\ \text{OMe} \end{array}$$

L4 ANSWER 14 OF 14 CAPLUS COPYRIGHT 2007 ACS on STN

AN 1973:478952 CAPLUS

DN 79:78952

TI [[(Trimethoxysilyl)propoxy]methyl]-1,3-dioxolanes

IN Koetzsch, Hans J.; Vahlensieck, Hans J.

PA Dynamit Nobel A.-G.

SO Ger. Offen., 11 pp.

CODEN: GWXXBX

DT Patent

LA German

FAN.CNT 1

TAN.CNI I						
PATENT NO.	KIND	DATE	APPLICATION NO.	DATE		
DE 2159991	A1	19730614	DE 1971-2159991	19711203		
DE 2159991	. C2 [·]	19821021				
GB 1364199	A	19740821	GB 1972-54097	19721122		
CH 575426	A5	19760514	CH 1972-17338	19721128		
US 3825567	Α	19740723	US 1972-310495	19721129		
FR 2162115	A1	19730713	FR 1972-42629	19721130		
JP 50040533	Α	19750414	JP 1972-121429	19721204		
US 3932464	Α	19760113	US 1974-469937	19740514		
US 4213908	A	19800722	US 1975-635533	19751126		
DE 1971-2159991	Α	19711203				
ÚS 1972-310495	A3	19721129	•			
US 1974-469937	A3	19740514	•			
	PATENT NO. DE 2159991 DE 2159991 GB 1364199 CH 575426 US 3825567 FR 2162115 JP 50040533 US 3932464 US 4213908 DE 1971-2159991 US 1972-310495	PATENT NO. KIND DE 2159991 A1 DE 2159991 C2 GB 1364199 A CH 575426 A5 US 3825567 A FR 2162115 A1 JP 50040533 A US 3932464 A US 4213908 A DE 1971-2159991 A US 1972-310495 A3	PATENT NO. KIND DATE DE 2159991 A1 19730614 DE 2159991 C2 19821021 GB 1364199 A 19740821 CH 575426 A5 19760514 US 3825567 A 19740723 FR 2162115 A1 19730713 JP 50040533 A 19750414 US 3932464 A 19760113 US 4213908 A 19800722 DE 1971-2159991 A 19711203 US 1972-310495 A3 19721129	PATENT NO. KIND DATE APPLICATION NO. DE 2159991 Al 19730614 DE 1971-2159991 DE 2159991 C2 19821021 GB 1364199 A 19740821 GB 1972-54097 CH 575426 A5 19760514 CH 1972-17338 US 3825567 A 19740723 US 1972-310495 FR 2162115 A1 19730713 FR 1972-42629 JP 50040533 A 19750414 JP 1972-121429 US 3932464 A 19760113 US 1974-469937 US 4213908 A 19800722 US 1975-635533 DE 1971-2159991 A 19711203 US 1972-310495 A3 19721129		

GI For diagram(s), see printed CA Issue.

AB The Si compds. I and II were prepared by reaction of (MeO)3SiH with the dioxolanes III (RR1 = O; R = R1 = H, resp.) over chloroplatinic acid in Me2CO at 70°. Refluxing I gave [3-(glycidyloxy)propyl]trimethoxysi lane.

IT 42345-73-3P 50650-15-2P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)

RN 42345-73-3 CAPLUS

CN 1,3-Dioxolan-2-one, 4-[[3-(trimethoxysily1)propoxy]methyl]- (CA INDEX NAME)

$$\begin{array}{c} \text{OMe} \\ \text{CH}_2\text{-O-(CH}_2)_3\text{-Si-OMe} \\ \text{OMe} \end{array}$$

RN 50650-15-2 CAPLUS

CN Silane, [3-(1,3-dioxolan-4-ylmethoxy)propyl]trimethoxy- (9CI) (CA INDEX NAME)